



Name: Chelsey Williams Date: 3/31/22

Student Exploration: Stoichiometry

Vocabulary: Avogadro's number, balanced equation, cancel, coefficient, conversion factor, dimensional analysis, formula mass, molar mass, mole, molecular mass, stoichiometry

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. A 250 mL glass of orange juice contains 22 grams of sugar. How much sugar is in a two-liter (2,000 mL) bottle of orange juice? 176 grams
2. It requires two sticks of butter to make a batch of 20 cookies. How much butter will it take to make 150 cookies? 15 sticks of butter

Gizmo Warm-up

Just as a cook follows a recipe to decide how much of each ingredient to add, a chemist uses **stoichiometry** to determine the amounts of substances involved in chemical reactions. The Stoichiometry Gizmo allows you to try your hand at figuring out the amounts of reactants and products that take part in a chemical reaction.

To begin, check that this equation is shown:



1. Look at the **coefficients** (such as the "3" in 3CO) in front of each substance in the equation. The coefficients tell you how many molecules or atoms take part in a chemical reaction. In the spaces below, list the number of each molecule or atom in the equation:

Fe_2O_3 1

CO 3

Fe 2

CO_2 3

2. In a **balanced equation**, the same number of each kind of atom is shown on each side of the equation. Calculate the number of iron (Fe), oxygen (O), and carbon atoms (C).

New equation



New question

How many moles of carbon monoxide (CO) are required to react completely with 1.75 moles of iron (III) oxide (Fe_2O_3)?

Problems solved: 0



Amadeo Creations Stoichiometry Answers

KJ Lindholm-Leary



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Table of Contents Amadeo Creations Stoichiometry Answers

1. Understanding the eBook Amadeo Creations Stoichiometry Answers
 - The Rise of Digital Reading Amadeo Creations Stoichiometry Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Amadeo Creations Stoichiometry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Amadeo Creations Stoichiometry Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Amadeo Creations Stoichiometry Answers
 - Personalized Recommendations

- Amadeo Creations Stoichiometry Answers User Reviews and Ratings
- Amadeo Creations Stoichiometry Answers and Bestseller Lists
- 5. Accessing Amadeo Creations Stoichiometry Answers Free and Paid eBooks
 - Amadeo Creations Stoichiometry Answers Public Domain eBooks
 - Amadeo Creations Stoichiometry Answers eBook Subscription Services
 - Amadeo Creations Stoichiometry Answers Budget-Friendly Options
- 6. Navigating Amadeo Creations Stoichiometry Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Amadeo Creations Stoichiometry Answers Compatibility with Devices
 - Amadeo Creations Stoichiometry Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Amadeo Creations Stoichiometry Answers
 - Highlighting and Note-Taking Amadeo Creations Stoichiometry Answers
 - Interactive Elements Amadeo Creations Stoichiometry Answers
- 8. Staying Engaged with Amadeo Creations Stoichiometry Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Amadeo Creations Stoichiometry Answers
- 9. Balancing eBooks and Physical Books Amadeo Creations Stoichiometry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Amadeo Creations Stoichiometry Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Amadeo Creations Stoichiometry Answers
 - Setting Reading Goals Amadeo Creations Stoichiometry Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Amadeo Creations Stoichiometry Answers
 - Fact-Checking eBook Content of Amadeo Creations Stoichiometry Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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