Balancing Chemical Equations Worksheet

1.
$$2 H_2 + 0_2 \rightarrow 2 H_2O$$

4.
$$2 N_2 + 0_2 \rightarrow 2 N_20$$

7.
$$Zn + Z HCI \rightarrow ZnCl_2 + H_2$$

8. ____ SiCl₄ + ___ H₂O
$$\Rightarrow$$
 ____ H₄SiO₄ + __ HCl

9.
$$2 \text{ Na} + 2 \text{ H}_2\text{O} \rightarrow 2 \text{ NaOH} + H_2$$

10.
$$_{-2}$$
 $H_3PO_4 \rightarrow _{---}$ $H_4P_2O_7 + _{---}$ H_2O

12. ____
$$CO_2$$
 + ___ Z $NH_3 \rightarrow ____ $OC(NH_2)_2$ + ____ $H_2O$$

13.
$$4 \text{ Si}_2\text{H}_3 + 7 \text{ O}_2 \rightarrow 8 \text{ SiO}_2 + 6 \text{ H}_2\text{O}_3$$

14.
$$2 \text{ Al}(OH)_3 + 3 \text{ H}_2SO_4 \rightarrow Al_2(SO_4)_3 + 6 \text{ H}_2O$$

16. ____ Fe₂(SO₄)₃ + ______ KOH
$$\rightarrow$$
 ______ K₂SO₄ + _____ Fe(OH)₃

17.
$$2 C_7H_6O_2 + 15 O_2 \rightarrow 14 CO_2 + 5 H_2O$$

19. ____ FeS₂ + ____ O₂
$$\rightarrow$$
 ____ Fe₂O₃ + ____ SO₂

22. ____ Na₂CO₃ + __2_ HCl
$$\rightarrow$$
 2 NaCl + ____ H₂O + ___ CO₂

24. ____
$$C_7H_{16} + _//_ O_2 \rightarrow _//_ CO_2 + _//__ H_{2}O$$

25. $P_4 + 5 O_2 \rightarrow 2 P_2O_5$

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