

Matter—Properties and Changes

Lesson 2.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical	mass	physical
density	properties	substance

Matter is anything with (1) _____ **MASS** _____ and volume. A (2) _____ **SUBSTANCE** _____ is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging (3) _____ **PROPERTIES** _____ that can be observed. Substances have both physical and chemical properties. (4) _____ **PHYSICAL** _____ properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) _____ **DENSITY** _____ are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) _____ **CHEMICAL** _____ properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either *physical* or *chemical*.

- | | |
|-------------------------|---|
| _____ CHEM _____ | 7. Chemical formula H_2O |
| _____ CHEM _____ | 8. Forms green carbonate when exposed to moist air |
| _____ CHEM _____ | 9. Remains unchanged when in the presence of nitrogen |
| _____ PHYS _____ | 10. Colorless |
| _____ PHYS _____ | 11. Solid at normal temperatures and pressures |
| _____ CHEM _____ | 12. Ability to combine with another substance |
| _____ PHYS _____ | 13. Melting point |
| _____ PHYS _____ | 14. Liquid at normal temperatures and pressures |
| _____ PHYS _____ | 15. Boiling point is $100^{\circ}C$ |
| _____ PHYS _____ | 16. Conducts electricity |

3 Properties Of Matter Study Guide

Carla C. Kirkland, Chan Cleveland



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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03
The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. Study Guide to Accompany Calculus for the Management, Life, and Social Sciences Clyde Metz, 1984-01-01
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