Name Class Date

Chapter 3 States of Matter

Section 3.2 The Gas Laws

(pages 75-91)

This section discusses gas pressure and the factors that affect it. It also explains the relationships between the temperature, software, and pressure of a new.

Reading Strategy (page 75)

Identifying Chase and Effect. As you need, identify the variables that affect gas pressure, and write them in the diagram below. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textwok.



Pressure tower 75-700

- What is presented? Entering it is result of a force distributed over an area.
- 2. Cityle the letter of each unit used to express amounts of pressure.

at registration.

b. Minde

(c.)panecal

- (rt.) (citerpasses)
- What causes the pressure in a closed container of gas? Colleges between periods of out and the walls of the container chain the pressure.

Factors that Affect Gas Pressure (sages 76-77)

4. Name the factors that affect the pressure of an enclosed gas.

a the temperature

the yeakperso

The resembles of the purfiction

- 8. Is the following sentence true or false? In a closed container, increasing the temperature of a gas will decrease the force with which particles bit the walls of the container.
- What effect does misting the temperature of a gas have on its pressure, if the volume of the gas and the number of its particles are kept constant? To presence of the gas off increase.
- 7. How does reducing the volume of a gas affect its pressure if the temperature of the gas and the number of particles are constant? Reducing the where of a gas recourse to pressure if the temperature and various are constant.

3 3 The Gas Laws Answer Key

Thomas Elliott Taylor

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