Name:

#### Chemistry Practice: Balancing Chemical Equations

 $OO_2$ 

- Balance the equations below. Remember this means only using coeeficients. All formulas are written correctly.
  - a.) 2 NaHCO<sub>3</sub> Na<sub>2</sub>CO<sub>3</sub> + H<sub>2</sub>O +
  - b.) Zn + S ZnS
  - c.) 2H<sub>2</sub> + O<sub>2</sub> → ZH<sub>2</sub>O
  - d.) 2 FeO + C ZFe + CO<sub>2</sub>
  - e.)  $Cl_2 + ZKI \longrightarrow ZKCI + I_2$
  - f.)  $B_2O_3 + 3Mg 3MgO + ZB$

  - (h) 5H<sub>2</sub>O + Na<sub>2</sub>B<sub>4</sub>O<sub>7</sub> + H<sub>2</sub>SO<sub>4</sub> → 4/H<sub>3</sub>BO<sub>3</sub> + Na<sub>2</sub>SO<sub>4</sub>
  - i.)  $CaC_2 + 2H_2O$   $\longrightarrow$   $Ca(OH)_2 + C_2H_2$
  - j.) NH<sub>4</sub>NO<sub>3</sub> N<sub>2</sub>O +2H<sub>2</sub>O
  - k.)  $H_2S + O_2$   $SO_2 + H_2O$
  - H<sub>2</sub>SO<sub>4</sub> + 2NaCl → ZHCl + Na<sub>2</sub>SO<sub>4</sub>

  - n.) 3 BaO + Z Al \_\_\_\_\_ 3 Ba + Al<sub>2</sub>O<sub>3</sub>
  - o.) 3 O<sub>2</sub> -> 2 O<sub>3</sub>
  - p.)  $SiO_2 + ZC$   $\longrightarrow$  ZCO + Si
  - q.) 2 PbS + 3 O<sub>2</sub> \_\_\_\_ 2 PbO + 2 SO<sub>2</sub>
  - r.) 2 HgO → 2 Hg + O<sub>2</sub>
  - s.)  $C_{12}H_{22}O_{11} + /ZO_2 /Z_CO_2 + //H_2O$
  - 1.)  $C_3H_8 + 5O_2 \rightarrow 3CO_2 + 4H_2O$

  - v.) P<sub>2</sub>O<sub>5</sub> + 6 NaOH 2 Na<sub>3</sub>PO<sub>4</sub> + 3 H<sub>2</sub>O
  - w.)  $3 \text{ BaCl}_2 + \text{Al}_2(SO_4)_3 \longrightarrow 3 \text{ BaSO}_4 + 2 \text{AlCl}_3$
  - x.) 4/NH<sub>3</sub> + 5 O<sub>2</sub> 6 H<sub>2</sub>O + 4/NO
  - y.) Z Na<sub>2</sub>O<sub>2</sub> + 2 H<sub>2</sub>O → 4/ NaOH + 4/ O<sub>2</sub>
  - z.)  $Sn(ClO_3)_4$   $\longrightarrow$   $SnCl_4 + 6 O_2$

# **10th Grade Chemistry Starr Practice Test**

**WJ Hussar** 

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web the developed design will be modeled using verilog hdl language which is a hardware description language hdl used to describe a digital system the verification of developed model will be made by identifying the suitable test cases in a test bench

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