

The Educare School Of Science & College

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Student Name:			Class:	Sr Year
Paper Type:	1A		Subject:	Chemistry
Paper Time:	3.00		Maximum Marks:	85

Q1. Choose the correct option.

(17x01=17)

1. One amu is equal to

- (A) 1.661×10^{-27} kg (B) 1.661×10^{-27} kg (C) 1.661×10^{-27} kg (D) 1.661×10^{-27} g

2. Tin has isotopes:

- (A) 7 (B) 9 (C) 11 (D) 5

3. The volume occupied by 1.4 g of O_2 at STP is:

- (A) 2.24 dm³ (B) 22.4 dm³ (C) 1.12 dm³ (D) 112 cm³

4. J. Berzelius determined the:

- (A) Atomic mass (B) Atomic numbers (C) Isotopes (D) Relative abundance

5. The largest number of molecules are present in:

- (A) 3.6 g of H_2O (B) 4.8 g of $\text{C}_2\text{H}_5\text{OH}$ (C) 2.8 g of CO (D) 5.4 g of N_2O_5

6. Which of the following element can exist in non-atomic form?

- (A) Oxygen (B) Chlorine (C) Nitrogen (D) Helium

7. A limiting reactant is the one which:

- (A) is taken in lesser quantity in gram as compared to other reactants (B) is taken in lesser quantity in volume as compared to other reactant (C) Give the minimum amount of the product which is required (D) Give the minimum amount of the product which under consideration

8. Ascorbic acid is vitamin:

- (A) A (B) B (C) C (D) D

9. In order to mention the boiling point of water at 110°C, the external pressure should be:

- (A) Between 200 torr and 760 torr (B) Between 760 torr and 1200 torr (C) 765 torr (D) Any value of pressure

10. In dry ice, CO_2 molecule form:

- (A) Ionic crystals (B) Covalent crystals (C) Molecular crystals (D) Any type of crystals

11. The carbon atom in diamond is:

- (A) sp^3 hybridization (B) sp -hybridization (C) sp^2 hybridization (D) dsp^2 hybridization

12. Transition temperature of tin(Sn) is:

- (A) 128°C (B) 95.5°C (C) 13.2°C (D) 32.8°C

13. Which of the following liquid has highest boiling point:

- (A) HCl (B) HBr (C) H_2O (D) Br_2

14. Dipole forces are present among:

- (A) Molecules of Iodine (B) Atoms of I_2 in vapor phase (C) Atoms of Neon in gases state (D) Molecules of chloroform

15. Dipole-dipole forces are present among:

- (A) Molecules of Iodine (B) Atoms of Iodine in vapor phase (C) Atoms of Neon in gaseous state (D) Molecules of chloroform

16. Vapor pressure of liquid depends upon:

- (A) Amount of liquid (B) Surface area (C) Temperature (D) Size of container

17. Transition temperature of FeWO_4

- (A) 13.2°C (B) 95.5°C (C) 128°C (D) 32.2°C

Q2. Write short answers of following questions.

(2x02=4)

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JS Bruner



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