

Stoichiometry Practice Problems

Name: _____

Date: _____ Period: _____

Reminder...Quiz next class on the following types of problems. Make sure you understand: 1. Molar mass 2. Mole Ratio 3. Avogadro's number.

Moles to Moles

1. Given the *unbalanced* equation: $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

a) If 4.50 moles of C_2H_6 are used, how many moles of oxygen are required?

b) If 0.3 moles of H_2O were produced, how many moles of carbon dioxide would also have been produced?

c) If you wanted to produce 5 moles of carbon dioxide, how many moles of oxygen would you needed to have started with?

Grams to Moles

2. Given the *unbalanced* equation below:



a) How many grams of H_2SO_4 are needed to react with 0.75 moles of NaOH?

b) How many grams of Na_2SO_4 would be formed if 0.75 moles of NaOH reacted?

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Huangqi Zhang



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