

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

1. Represent a chemical reaction equation
2. Balance a chemical reaction equation
3. Write chemical formulas for ionic compounds (homopolyatomic)

Vocabulary

- | | |
|----------------------|----------------------|
| 1. chemical equation | 4. coefficient |
| 2. homopolyatomic | 5. balanced equation |
| 3. ionic | |

Part A Completion

Use the vocabulary words in each sentence, finding either a synonym or an antonym. Write the word in the space provided.

1. A chemical reaction is a process in which _____.
2. The chemical formula of a chemical compound is the _____ the total number of atoms of each element in the _____.
3. The law of conservation of _____ is demonstrated by the fact that the number of _____ is the same on both sides of a balanced equation.
4. _____ is a chemical reaction in which a substance reacts with oxygen to form a compound.
5. _____ is a chemical reaction in which a substance reacts with a substance to form a compound.

6. The chemical formula of a compound is _____.
7. The chemical formula of a compound is _____.
8. The chemical formula of a compound is _____.
9. The chemical formula of a compound is _____.
10. The chemical formula of a compound is _____.
11. The chemical formula of a compound is _____.
12. The chemical formula of a compound is _____.
13. The chemical formula of a compound is _____.
14. The chemical formula of a compound is _____.
15. The chemical formula of a compound is _____.

1. equation
2. balanced
3. coefficient
4. ionic
5. homopolyatomic
6. balanced
7. coefficient
8. equation
9. balanced
10. coefficient
11. equation
12. balanced
13. coefficient
14. equation
15. balanced

11 1 Describing Chemical Reactions Section Review

Answer Key

Linda Swisher, Kevin T. Patton



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