# Chemical Reactions

### Signs of Chemical Reactions

What is a chemical reaction?

### Chemical Formulas

H<sub>2</sub>O weater

A chamical formula uses \_ too iddentify the in a compound and the

of each kind present in a molecule.

# C6H12O6 glucose

A glucese molecule is exade up of \_\_\_\_

# **Chemical Equations**

\* attorno

A chamical equation uses to describe a

chemical reaction. The starting materials in a chemical reaction are . The resulting substances from a reaction

placed on the left of the arrow, and the the right of the arrow.

C+O, - CO.

### Types of Chemical Reactions

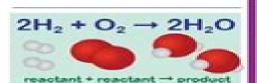
### Synthesis Reaction

Describe a synthesis reaction:

A warter mobile-pulse is made

bonded to \_\_\_\_\_ atom of

up-of \_\_\_\_\_ atoms of





contribution to Foreign in more abject.

### Decomposition Reaction

attorns of

Describe a decomposition reaction:

2H₂O -	• 2H <sub>2</sub> + O <sub>2</sub>
reactant	product + product
1	

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# **Key Vocabulary Terms**

- double replacement reaction.
- charmical equation. \* effectivene chemical formula \* controller to
  - \* PRINCESOR
- chemical reaction. · compound
  - single replacement reaction
- decomposition reaction swithesis reaction.

### Replacement Reaction

#### Single Replacement

Describe a single replacement reaction:

Zn + 2HCl → ZnCl<sub>2</sub>+ H<sub>2</sub> reactant + reactant -+ product + product



#### Double Replacement

Describe a double replacement reaction:

NaCl + AgF → NaF + AgCl reactant + reactant -+ product + product

# **11 Chemical Reactions Guide Key**

**Geoff Neuss** 

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