## **Chemical Reactions**

#### Part A. Vocabulary Review

Directions: Complete the following sentences using the terms listed below.

- synthesis exothermic catalyst decomposition coefficients endothermic single-displacement inhibitor reactants precipitate chemical double-displacement products balanced L. A change in which one or more substances are converted into a new substance is called a(n) double-displacement reaction.
- If a precipitate, water, or a gas forms when two ionic compounds in solution are combined, the reaction is a(n) precipitate reaction.

3.A substance that prevents certain chemical reactions is a(n) inhibitor\_

- When two or more substances combine in a chemical reaction to form another substance, a(n) synthesis reaction has taken place.
- The breakdown of a substance into two or more simpler substances is a(n) decomposition reaction.
- In a chemical equation, the substances that react are called reactants
- A substance that speeds up a chemical reaction without itself being permanently changed is a(n) catalyst
- 8. A reaction that requires energy in the form of heat is a(n) endothermic reaction.
- When one element replaces another in a compound, a(n) single-displacement reaction has occurred.
- If the energy released in a chemical reaction is primarily in the form of heat, that reaction is a(n) exothermic reaction.
- Numbers in front of the symbols and formulas in an equation are called coefficients
- 12. If the number of atoms of each element on the left side of an equation is equal to the number of atoms of each element on the right side of the equation, the equation is **balanced**
- 13. In a chemical equation, the substances that are produced are called products
- 14. An insoluble solid that forms when two ionic compounds in solution are combined is
- a(n) chemical reaction

# 11 Chemical Reactions Vocabulary Review Answer Key

**Morris Hein, Susan Arena** 

#### 11 Chemical Reactions Vocabulary Review Answer Key:

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