Second Terminal Examination (2022-23)

Subject- CHEMISTRY

Time; - 3 Hrs.

Class- 11th Science

Markst - 76

Instructions:

Question paper consist of 31 questions aivided into four sections namely A. B. C. D.

- 1) Section-A. Q. No. I contains ten multiple choice types of questions carrying one mark each. Q.No.2 contains eight very short answer types of questions carrying one mark each.
- 2) Section-B : Q.No.3 to Q. No.14 are Twelve questions carrying two mark each out of which any eight to be attempted.
- 3) Section-C : Q. No.15 to Q.No.25 are Twelve questions carrying three mark each out of which any eight are to be attempted.
- 4) Section-D: Q No.27 to Q No. 31 are Five questions carrying four mark each out of which any three to be attempted

Section-A

().1) Select and write the correct answer:

43.00%

- Which of the following compounds can NOT demonstrate the low of proportions?
 - a) NO.NO
- b) CO,CO;
- c) H-O.H-O.
- d) Na-S, NaF
- Which of the following temperature will read the same value on Celsius and Fabrenbeit Scales
 - 23 -40

- b) = 40
- C1 -SD

- d1 -20
- SI unit of the quantity electric current is b) Ampere e) Candela
 - a) Volts

- di Newton
- The no. of molecules in 22.4 cm3 of Nitrogen Gas at STP is
 - a) 6.022 10²⁰ b) 6.022 10²³ c) 22.4 + 10²⁰
- d) 22.4 1023
- The homogeneous series of alcohol has general molecular formula? 1970
 - a) Cn H₂n+₂OH b) Cn H₂n OH
- c) Cn Han+ OH

- Find the molar mass of CH₃ CHO.
 - 0) 44

b) 40

CINO

d) 30

- Find the molar mass of CH₁ NO: WHILE.
 - 6) 77

c) 98

d) 51

- Atomic number of an elements is equal to
 - 11 P c-

b) p-N

CIPSA

dr p = n = e

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