

Q 3. Answer the following questions.

- (i) Discuss law of Reciprocal Proportion with examples.
- (ii) An inorganic salt gave the following percentage composition:
 $\text{Na} = 29.11\%$, $\text{S} = 40.51\%$ & $\text{O} = 30.38\%$, Calculate the Empirical Formula of the salt.
- (iii) Discuss Aufbau's Principle and Hund's Rule.
- (iv) What is diagonal relationship with example.
- (v) Explain sp^2 hybridization with example.
- (vi) On the basis of VSEPR theory, predict the shapes of the following molecules/ions?

(a) SF_4	(b) NH_4^+	(c) H_3O^+
(d) NH_2	(e) PCl_5	(f) NH_3
- (vii) State & explain Hess's Law of constant heat summation.
- (viii) What is Buffer Solution. Give its types.
- (ix) Give three methods of preparation of Alkynes.

SECTION D

Q 4. Answer the following questions.

- (i) What are Quantum numbers? Explain
Or
State and Explain Heisenberg's Uncertainty Principle.
- (ii) Define 1st Law of Thermodynamics. Derive its mathematical formulation.
Or
Define Heat Capacity. Derive relation between C_p & C_v .
- (iii) Explain Nucleophilic Substitution Reaction.
Or
Write the IUPAC names of the following organic compounds:

(a) $\text{CH}_3 - \text{CH}_2 - \text{CH} = \text{CH} - \text{CHO}$
(b) $\begin{array}{c} \text{COOH} \\ \\ \text{COOH} \end{array}$

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TD Snyder



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11th Question Paper Chemistry 2013 Book Review: Unveiling the Power of Words

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