(i) Donie bond (ii) Covalent bond (iii) Co-ordinate bond

In addition to these 3 main types of Chemical bonds, the atoms in a metal are linked by metallic bonds while hydrogen bond and Vander waals forces arise because of intermelecular forces of attraction in the covalent mode - cules.

* TONIC BOND OR ELECTROVALENT BOND :-

The bond formed by complete transference of E from one atom to another so that both of them acquire nearest nable gas configuration is called <u>ionic bond</u>. It is formed by atoms having large electronegativity difference. The atom which loses E acquire the charge the charge former is the thought of the theory of the strong electrostatic force of attraction.

Examples :-

1. formation of Nacl: - Sodium atom (Z=11) has one valence E (2,8,1). IIIly chlorine atom (Z=17) has 7

Valence Es (2,8,7). Sodium atom loses its only valence is and changes Na+ion (cotion). This E is taken up by chlorine atom which changes to (I ion (anion). But these ions are mutually attracted to form Nacl as shown below:

$$Na \times + : \stackrel{\cdot}{\leftarrow} : \longrightarrow [Na]^{+} [\stackrel{\cdot}{\sim} : \stackrel{\cdot}{\leftarrow} :]$$
 $(2,8,1) \quad (2,8,7) \quad (2,8) \quad (2,8,8)$

(caf2):-

@ formation of magnesium oxide (Mgo) :-

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