

12.3 Limiting Reagent and Percent Yield >

Limiting and Excess Reagents

Experimental Conditions		
	Reactants	Products
Before reaction	 2 molecules N_2  3 molecules H_2	0 molecules NH_3
After reaction	 1 molecule N_2  0 molecules H_2	 2 molecules NH_3

- In this reaction, only the hydrogen is completely used up.
- H_2 is the **limiting reagent**, or the reactant that determines the amount of product that can be formed by a reaction.

12 3 Limiting Reagent And Percent Yield Stoichiometry

**Allan Blackman, Daniel
Southam, Gwendolyn Lawrie, Natalie
Williamson, Christopher Thompson**

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