

Chapter 12

Section review 1

3. a

4. no; Temperature is the average kinetic energy of particles of a substance. If the individual particles of two objects have the same average kinetic energy, they are at the same temperature regardless of how much matter is in each object.

5. When the temperature of an object is raised, its particles move around more, so they take up more space. This causes the entire object to expand.

$$6. (9/5 \times 35^{\circ}\text{C}) + 32 = 95^{\circ}\text{F}$$

$$7. 5/9 \times (34^{\circ}\text{F} - 32) = 1.1^{\circ}\text{C}$$

$$8. 0^{\circ}\text{C} + 273 = 273 \text{ K}$$

$$9. 100 \text{ K} - 273 = -173^{\circ}\text{C}$$

10. The soup could overflow its pot as it cooks because of thermal expansion. The soup will expand in volume as its temperature increases. If the cold soup is too close to the top of the pot, it will likely overflow as it expands.

11. Thermal expansion causes the density of a substance to decrease, because the same mass takes up more volume.

12. Thermal energy was transferred from the table to the glass of cold water.

Section review 2

3. c

4. Heat is the transfer of thermal energy between objects at different temperatures.

$$5. 128 \text{ J/kg} \cdot ^{\circ}\text{C} \times 0.015 \text{ kg} \times 10^{\circ}\text{C} = 19 \text{ J}$$

6. no; They will not be at the same temperature, because in order for a larger object to have the same total thermal energy as a smaller object, the larger object must be at a lower temperature.

7. Because wood is a poor conductor of heat, you can hold the wooden handle of a hot pot, for instance, without being burned.

8. a: conduction; b: convection; c: radiation

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Albert John Walford



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J. G. R. Briggs,2009 Energy Research Abstracts ,1987 Chemistry insights 'O' level Rex M. Heyworth,2007

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