

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	Aluminum
_____	14-18
_____	Transition?

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It exists in the periodic table.
- _____ 20. It has no line periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are strong oxidizing agents.
- _____ 23. Group 18 elements.

Column B

- a. noble
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 24. There are two representative elements of elements.
- _____ 25. Most elements have a line of the periodic table are noble.
- _____ 26. Group 1 elements except for hydrogen are found in the alkali metals.
- _____ 27. Group 18 elements are the alkaline earth metals.
- _____ 28. Group 1 elements are highly reactive elements known as alkalis.
- _____ 29. Group 18 elements are very reactive elements known as noble gases.
- _____ 30. Alkaline earth metals are found in the periodic table.

Answer Key To Study Guide For Content Mastery Chapter 1

Leslie S Treas, Judith M Wilkinson



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