

UNDERSTANDING THE MOLE AND AVOGADRO'S NUMBER

Name _____	Section _____	Date _____
Understanding the Mole and Avogadro's Number Lab		
One mole of anything is a very large number of things. It is difficult to understand just how large a mole is. This lab serves as a way to understand how truly large a mole of things is.		
Your teacher will give you one or more small items and pick one or more places. Your goals for this lab are:		
1. Calculate the volume of the item or items that your teacher gives you. You may measure the volume of one item or measure the volume of several of the items using a technique like water displacement and then calculate the volume of a single item through mathematics. 2. Calculate the volume of one mole of that item. 3. Calculate the volume of the place or places that your teacher gives you. 4. Compare the volume of one mole of the item to the volume of the place. Which one is larger? If the volume of the place is larger, one mole of the item will fit. If the volume of one mole of the item is larger, then one mole of the item will not fit in the place.		
For example: you may be given an eraser as your object and your classroom as a place.		
For one mole, use Avogadro's number: _____		

Name _____
Calculations and the Mole
• Item ____ is a _____ • The volume of this item is _____ • The volume of one mole of this item is _____ • Place ____ is the _____ • The volume of this place is _____
• Will one mole of item 1 fit in this place? Circle one Yes No • If no, how many of this place will be needed to hold one mole of the item? _____

A CHEMISTRY LAB
10 PROBLEMS
3 VERSIONS
WORKED ANSWER KEY

Answer Key To Understanding The Mole Lab

Daniel F McAuley



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Energy Research Abstracts, 1992-05 *The Extraordinary Chemistry of Ordinary Things, Study Guide* Carl H. Snyder, David F.

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