

Question 1

1(a)(iii) Difference: With powdered CaCO_3 the graph has a steeper slope at the start of the reaction than with marble chips because the rate of reaction is faster due to increased surface area of the solid.
Similarity: The total volume of CO_2 produced is the same in both cases since the same mass (and therefore number of moles) of CaCO_3 was used in both experiments.



1(a)(v) Moles of CaCO_3 used = $(1.8/100)$ mol. = 0.018 mol.
From the balanced equation for the reaction, 1 mole CaCO_3 produces 1 mol CO_2 .
Hence volume of gas produced at R.T.P. = (0.018×24000) cm^3 = 432 cm^3 .

1(a)(vi) There were impurities in the CaCO_3 used.

1(a)(vii) Concentration of the acid used and temperature at which the reaction was carried out.

1(b)(i) $\text{Fe}^{3+}_{(\text{aq})}$ present.

1(b)(ii) Magnesium ribbon decreases in size/dissolves.
Solution becomes pale green.



1(b)(iv) SO_4^{2-} , SO_3^{2-} and CO_3^{2-} absent.

1(b)(v) White precipitate, which is soluble in aqueous ammonia, is formed.

Answers For January 2014 Chemistry

S Marginson



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international journals He has more than four years of teaching experience in various engineering colleges

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importance of adopting greener practices Environmentalists have long raised alarms about these invisible killers prompting governments and chemists to seek greener solutions This book explores the development of environmentally friendly products such as toiletries and detergents replacing harmful substances with safer alternatives We discuss the pervasive issue of plastic pollution and its devastating effects on ecosystems Awareness and education are crucial in promoting sustainable practices and this book aims to enlighten readers on the importance of green chemistry Chemists are leading the way by utilizing renewable resources contributing to a healthier and more sustainable world *The Air Force Law Review*

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