

8. Buffer, passing question 10, is shown within the Earth's upper crust. Buffer is composed of the components of those of ascending velocity.

A sample of Buffer from a sample was analyzed to give the following composition of samples.

Sample	Composition (%)
1%	0.001
1%	0.01
1%	0.02

- (a) Define the term relative atomic mass.

The relative atomic mass of an element is the average mass of the atoms of the element, taking into account the relative abundance of the isotopes of the element.

- (b) Calculate the relative atomic mass of the sample of buffer.

Give your answer to three decimal places.

$$\left(\frac{1}{100} \times 1.001 \right) + \left(\frac{1}{100} \times 1.01 \right) + \left(\frac{1}{100} \times 1.02 \right) = 1.011$$

Answer = 1.011 ± 0.001

- (c) Some studies, as early as 1950, suggested that specimens were made up of two particles which could be the subatomic particles, electrons and protons.

Complete the table to show the number of subatomic particles in the ${}^1\text{H}$ atom and ${}^2\text{H}$ ion.

	protons	neutrons	electrons
${}^1\text{H}$	1	0	1
${}^2\text{H}^+$	1	1	0



Answer To F321 Chemistry May Series

JR Anderson



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