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Exercise 6.39: Problems - The Mole Concept

Part A

How many carbon atoms are in a diamond (pure carbon) with a mass of 75 mg ?

Express your answer using two significant figures.

$N =$ $2.5 \cdot 10^{21}$ C atoms

You have already submitted this answer. Enter a new answer.
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Answers Mastering Chemistry Homework

Roscoe, Rod D.,Craig, Scotty D.,Douglas, Ian

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