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and may come with a short warranty or none at all In addition used cars may lack the latest safety features That is why it is so important for consumers to do extensive research so they can avoid all of the potential pitfalls of buying a used car The auto experts at Consumer Reports have done the work for you and have compiled their extensive research and report their findings into the 2007 edition of USED CAR BUYING GUIDE This fabulous tool will help steer any consumer who is in the market for a used car towards the better performing and more reliable used car models and away from those models with a troubled past or substandard performance Before consumers set foot on a used car lot they should read all the valuable information provided in this book so they can be armed with as much information as possible and the knowledge to make an educated choice Consumer Reports knows cars and offers the most detailed and revealing used car reliability information available anywhere including Unbiased reviews of every major model from 1999 2006 Lists of the best and worst used vehicles and how to avoid a lemon A checklist of what to look for when inspecting a used car Best used cars for gas mileage Tips on negotiating the best price Reliability recalls and crash test information Making sense of safety information How to get the most money when trading in your current car The majority of this book is devoted to the profiles of 264 cars minivans SUVs and trucks presenting all major 1999 2006 models Each profile contains a photo from the representative year a write up of the vehicle reliability history crash test data and the model years when key safety gear was added and when a major redesign was made

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web jun 14 2015 this list of electron configurations of elements contains all the elements in increasing order of atomic number to save room the configurations are in noble gas shorthand this means part of the electron configuration has been replaced with the element symbol of the noble gas symbol

web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of 1 s 2 2 s 1

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the

atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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