



QUIZ: THE PROPERTIES OF MATTER

1. Match the following descriptions with the property of matter that best fits the description. Each term may be used once, more than once, or not at all. (25)

<u>5</u>	a) Lemons are more sour than cherries.	A) boiling point
<u>8</u>	b) It is easier to see the view through an ordinary window than through a stained glass window.	B) clarity
<u>M</u>	c) Ice begins to form water at 0°C.	C) color
<u>10</u>	d) Copper can be stretched into thin wires and is often used to make piping and electrical wiring.	D) combustibility
<u>1</u>	e) At room temperature, mercury is the only metal found as a liquid.	E) crystal form
<u>12</u>	f) After painting with oil paints, the oil cannot be removed by water, but it can with turpentine.	F) density
<u>6</u>	g) Dry wood catches fire more easily than damp wood.	G) ductility
<u>18</u>	h) When sodium is placed in water, it produces a lot of heat and catches on fire.	H) electrical conductivity
<u>1</u>	i) Aluminum foil is easily shaped into thin sheets.	I) hardness
<u>1</u>	j) My cat's hair is soft, but my dog's hair is stiff.	J) luster
<u>12</u>	k) Sugar can dissolve in water when the two substances are mixed.	K) magnetism
<u>1</u>	l) When using HB lead in a mechanical pencil, it breaks more easily than using H lead.	L) malleability
<u>12</u>	m) It takes longer for honey to pour out of a container than it takes for water.	M) melting point
<u>8</u>	n) A sample of lead and a sample of aluminum have the same volume, but lead has the greater mass.	N) odor
<u>1</u>	o) Using a magnifying glass, you can see that salt forms small cubes, but baking soda is a powder.	O) reactivity with acid
		P) reactivity with water
		Q) solubility
		R) state
		S) taste
		T) texture
		U) viscosity

2. Describe one difference between a physical property and a chemical property. (2)

A physical property is one that can be observed or measured without changing the identity of the substance, while a chemical property is observed or measured while a substance is interacting with another substance, which results in a change of its identity.

3. Define the following properties of matter. (3)

- A) **luster**
This property of a substance determines how much light reflects off of its surface.
- B) **electrical conductivity**
This property of a substance determines if it will allow an electrical current to run through it or not.
- C) **boiling point**
This property of a substance is the temperature at which it will turn from its liquid state to its gas state.

Properties Of Matter Wordwise Answers

Gordon W. Brown



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