Oxidation-Reduction Worksheet

For each reaction below, identify the atom oxidized, the atom reduced, the oxidizing agent, the reducing agent, the oxidation half reaction, the reduction half reaction, and then balance the equation by the method of oxidationreduction showing all electrons transfers.

5.
$$KMnO_4 + NaCl + H_2SO_4 à Cl_2 + K_2SO_4 + MnSO_4 + H_2O + Na_2SO_4$$

8.
$$H_2C_2O_4 + K_2MnO_4 à CO_2 + K_2O + Mn_2O_3 + H_2O$$

9.
$$Mn(NO_3)_2 + NaBiO_3 + HNO_3 à HMnO_4 + Bi(NO_3)_3 + NaNO_3 + H_2O$$

Chapter 20 Oxidation Reduction Reactions Worksheet

Chris Hren, Peter J. Mikulecky

Chapter 20 Oxidation Reduction Reactions Worksheet:

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