

Name: _____

Date: _____

Reading the Periodic Table **KEY**

1. List the atomic numbers of the elements in Period 2.

3, 4, 5, 6, 7, 8, 9, 10

2. List the symbols of the elements in Period 3.

Na, Mg, Al, Si, P, S, Cl, Ar

3. Name the elements in Period 2 that are metals.

Li, Be

4. Name the elements in Period 1 that are nonmetals.

H, He

5. Which element in Period 3 has the least mass?

Sodium (Na)

6. Which element in Period 4 has the most mass?

Krypton (Kr)

7. To which group/family do these belong?

a. Cobalt **transition metals**

b. Sodium **alkali metals**

c. Argon **noble gases**

d. Silicon **group 14 metalloids**

e. Chlorine **halogens**

f. Magnesium **alkaline earth metals**

8. Find iron on the periodic table:

a. What is the atomic number of iron? **26**

b. To which group does iron belong? **Transition metals**

9. Name the elements that have properties similar to that of chlorine.

Fluorine, bromine, iodine, astatine, tennessine

10. Fill in the chart below. The first row has been completed as an example.

Element	Symbol	Atomic Number	Period	Group
sodium	Na	11	3	1
oxygen	O	8	2	16
krypton	Kr	36	4	18
gold	Au	79	6	11
cadmium	Cd	48	5	12
manganese	Mn	25	4	7

Exploring The Periodic Table Answer Key

Cindy Blobaum



Exploring The Periodic Table Answer Key:

Exploring Earth and Space Michael DiSpezio,1995 A textbook exploring such aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter **Periodic Table** Cindy Blobaum,2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket **Exploring Various Side**

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