Excretion

CHEMICAL REACTIONS AND

| | AUS | HONS |
|--|--|------------------|
| Objection | ve Questions | |
| Multiple Choice Questions | Questions | |
| 1. Which of the | | |
| 1. Which of the statements about to (i) Lead is only 2PbO(s) + C | he reaction below | |
| (i) Lead is getting reduced | $(s) \longrightarrow 2Pb(S) + CO_2(2)$ | |
| (a) Carbon dioxide in | 22 | |
| (iii) Carbon is getting oxidised | Tised | |
| The state of the s | | |
| 7. 1.7 | (b) (i) and (iii) | |
| (c) (i), (ii) and (iii) | (d) all of these. | |
| 2. Fe ₂ O ₃ + 2AI - Al ₂ O ₃ + 2Fe The above reaction is an example | | (2924) |
| (a) combination reaction | | |
| (c) decomposition reaction | (b) double displacement rea | ction |
| . What happens when dilute byde- | (d) displacement reaction. | |
| (a) Hydrogen gas and iron chlorid | le are produced | ings ? (2022) |
| () cinornie gas and fron hydroxi | de are produced | |
| (c) 140 reaction takes place | | |
| (d) Iron salt and water are produce | od. | |
| In the reaction 2FeCl ₂ + Cl ₂ | 2FeCl ₃ Chlorine is a- | |
| (a) Reducing agent | (b) Oxidizing agent | |
| (c) Catalyst | (d) Inert medium. | |
| Rancidity can be prevented in foo | | |
| (a) Mixing antioxidant | (b) Keeping in refrigerator | |
| c) Keeping away from light | (d) All of these. | |
| Which among the following is not | | |
| a) Oxygen | (b) Cone. Sulphuric acid | |
| c) Chlorine | (d) Hydrogen. | |
| Vhich among the following gases | | items fresh ? |
| Carbon dioxide and Oxygen | (b) Nitrogen and helium | |
|) Helium and Oxygen | (d) Nitrogen and Oxygen. | |
| hemically rust is— | Chile all side on the State of | |
|) Hydrated ferrous oxide | (b) Ferric oxide | |
|) Hydrated ferric oxide | (d) None of these. | |
| exothermic reaction in which | clucose combines with oxyger | a in the cens of |
| r body and provides energy- | | |
| Respiration | (b) Digestion | |
| Excretion | (d) None of these. | |

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John Wilkinson

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