Types of Chemical Reactions - Guided notes

Types of Chemical Reactions:

1 Synthesis

2 Decomposition

3 Single Replacement

4 Double Replacement

5 Combustion

Synthesis Reaction: Two or more substances combine to form a single substance.

General equation:

A + B - AB

Example:

2 Na + Cl₂ → 2NaCl

Decomposition Reaction: A compound is broken into two or more products,

General equation:

AB - A + B

Example:

2 H₂O → 2H₂ + O₂

Single Replacement: when one element replaces another in a compound,

General equation:

 $A + BC \rightarrow AC + B$

Example:

Mg + CuCl₂ → MgCl₂ + Cu

Double Replacement: occur when a metal replaces a metal in a compound and a nonmetal replaces a nonmetal in a compound

General equation:

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Clyde Metz

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