

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

- 1. Represent the processes of a reaction.
- 2. Explain how to write a balanced equation.
- 3. List the steps for writing a complete ionic equation.

Vocabulary

- 1. chemical equation
- 2. balanced equation
- 3. catalyst
- 4. coefficient
- 5. skeletal equation

Part A Completion

Write the coefficient for each in these chemical equations, and balance and label every chemical reaction. Indicate the physical states of the reactants and products.

1. A chemical equation is always represented by a balanced _____ 1 _____
 The chemical formula for a chemical compound is _____ 2 _____ The mass
 number for hydrogen is _____ 3 _____ The mass number for
 the new compound is _____ 4 _____ A chemical equation is a reaction that
 represents a chemical reaction. The _____ 5 _____ is the chemical equation
 for the reaction. The chemical equation is balanced by changing the _____ 6 _____ of
 the chemical formula for the reactants.

2. The chemical equation for the reaction of hydrogen and oxygen is a
 balanced chemical equation. The chemical equation is _____ 7 _____ that a catalyst _____ 8 _____ is used for a reaction.
 _____ 9 _____ A chemical equation is a reaction that is balanced by changing the _____ 10 _____ of the
 chemical formula for the reactants and products.

- 1. _____
- 2. _____
- 3. _____
- 4. _____
- 5. _____
- 6. _____
- 7. _____
- 8. _____
- 9. _____
- 10. _____

1describing Chemical Reactions Section Review Answers

Kevin T. Patton, Gary A. Thibodeau



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