

Unit 5 – Chemical Reactions

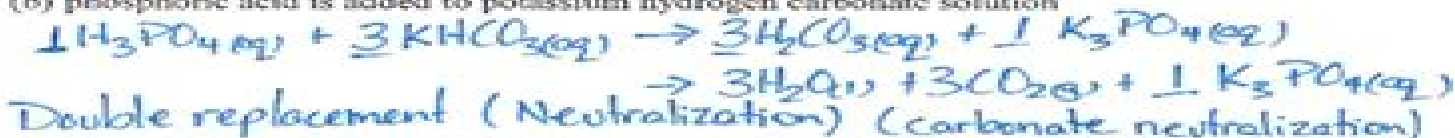
13 Write complete balanced chemical equations for the following. Classify each reaction as combination, decomposition, single replacement, double replacement, neutralization, or combustion. Don't forget your subscripts!

(a) potassium reacts with bromine



Combination

(b) phosphoric acid is added to potassium hydrogen carbonate solution



Double replacement (Neutralization) (carbonate neutralization)

(c) magnesium is added to nickel (II) nitrate solution



Single Replacement.

(d) hydrochloric acid is added to ammonium hydroxide.



Double Replacement (neutralization)

(e) nickel is added to magnesium chloride solution



(f) copper (II) sulphate pentahydrate is produced in a combination reaction



Combination

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