

2.2 Properties of Water

Lesson Objectives

-  Discuss the unique properties of water.
-  Differentiate between solutions and suspensions.
-  Explain what acidic solutions and basic solutions are.

Lesson Summary

The Water Molecule Water molecules (H_2O) are polar because of an uneven distribution of electrons, creating a slight negative (–) charge in the oxygen atom and a slight positive (+) charge in each hydrogen atom. The attraction between a hydrogen atom of one water molecule and the oxygen atom of another water molecule is called a **hydrogen bond**.

- ▶ **Cohesion** is an attraction between molecules of the same substance. It causes water molecules to be drawn together, producing surface tension.
- ▶ **Adhesion** is an attraction between molecules of different substances. It causes capillary action, an effect that causes water to rise in a narrow tube against the force of gravity.

Solutions and Suspensions A **mixture** is a material composed of two or more elements or compounds that are physically mixed together but not chemically combined. A **solution** is a mixture in which all the components are evenly spread out: the substance dissolved is the **solute**; the substance that causes the dissolving is the **solvent**. Mixtures of water and undissolved materials are **suspensions**.

Acids, Bases, and pH A water molecule (H_2O) can split apart to form a hydrogen ion (H^+) and a hydroxide ion (OH^-).

- ▶ The **pH scale** measures the concentration of hydrogen ions in a solution. The scale ranges from 0 to 14. Pure water has a pH of 7.
- ▶ An **acid** is any compound that forms H^+ ions in solution. Acidic solutions have pH values below 7. A **base** is a compound that forms OH^- ions in solution. Basic, or alkaline, solutions have pH values above 7.
- ▶ **Buffers** are weak acids or bases that can react with strong acids or bases to prevent sudden changes in pH.

The Water Molecule

For Questions 1–4, write True or False on the line provided.

- _____ 1. Water is a polar molecule.
- _____ 2. Hydrogen bonds are an example of adhesion.
- _____ 3. Covalent bonds give water a low heat capacity.
- _____ 4. A hydrogen bond is stronger than a covalent bond.

2 2 Properties Of Water Answers

**Michael Grantham,P A Muhammed
Basheer,Bryan Magee,Marios Soutsos**

2 2 Properties Of Water Answers:

Properties of Water in Foods D. Simatos, J.L. Multon, 2012-12-06 Water is recognized as being an important factor in numerous phenomena connected with the quality of food. For instance, it plays a part in the textural properties of several commodities. Moreover, water is an essential parameter determining the behaviour of food products in the course of many processing operations. On water, will depend the amount of energy necessary for freezing or dehydrating the product. Water will strongly influence the evolution of physical, chemical and biochemical phenomena taking place in the product during processing operations such as heating, drying, etc. Water will also influence the same reactions as well as the activity of microorganisms during the storage of food products under various conditions. As a result, all aspects of quality, sensory, nutritional and hygienic properties of the food will be affected. In all these circumstances, the water content of a product is obviously an important factor, but equally important may be the physical properties of this water, such as its thermodynamic activity and its mobility. Actually, the concept of water activity is now widely used by the food industry and in the legislation of several countries. The idea of a small international meeting devoted to a synthetic review and discussion of knowledge on these various matters was first developed by Dr R B. *Nuclear Science Abstracts*, 1970. NSA is a comprehensive collection of international nuclear science and technology literature for the period 1948 through 1976, pre-dating the prestigious INIS database which began in 1970. NSA existed as a printed product, Volumes 1-33, initially created by DOE's predecessor, the U.S. Atomic Energy Commission (AEC). NSA includes citations to scientific and technical reports from the AEC, the U.S. Energy Research and Development Administration and its contractors, plus other agencies and international organizations, universities and industrial and research organizations. References to books, conference proceedings, papers, patents, dissertations, engineering drawings and journal articles from worldwide sources are also included. Abstracts and full text are provided if available. *Ebook: Chemistry: The Molecular Nature of Matter and Change* Silberberg, 2015-01-16. *Ebook: Chemistry: The Molecular Nature of Matter and Change - Molecular Theory Of Water And Aqueous Solutions - Part 1: Understanding Water* Arieh Ben-naim, 2009-07-17. The aim of this book is to explain the unusual properties of both pure liquid water and simple aqueous solutions in terms of the properties of single molecules and interactions among small numbers of water molecules. It is mostly the result of the author's own research spanning over 40 years in the field of aqueous solutions. An understanding of the properties of liquid water is a prelude to the understanding of the role of water in biological systems and for the evolution of life. The book is targeted at anyone who is interested in the outstanding properties of water and its role in biological systems. It is addressed to both students and researchers in chemistry, physics and biology. *Fluctuation Theory of Solutions* Paul E. Smith, Enrico Matteoli, John P. O'Connell, 2016-04-19. There are essentially two theories of solutions that can be considered exact: the McMillan-Mayer theory and Fluctuation Solution Theory (FST). The first is mostly limited to solutes at low concentrations, while FST has no such issue. It is an exact theory that can be applied to any stable

solution regardless of the number of components and their co

Water: A Comprehensive Treatise Felix Franks, 2012-12-06 Since the publication of the previous volumes many new aspects of the physical and life sciences have been developed in which the properties of water play a dominant role Although according to its preface Volume 5 was to be the last one of the treatise these recent developments have led to a revision of that statement The present volume and its companion still in preparation deal with topics that were already mentioned in the preface to Volume 5 as gaining in importance The recent development of X ray and more particularly neutron scattering techniques have led to studies of structure in aqueous solutions of electrolytes on the one hand and to the role of water in protein structure and function on the other Both these topics have reached a stage where reviews of the present state of knowledge are useful The application of ab initio methods to calculations of hydration and conformation of small molecules has a longer history but here again a critical summary is timely The role of solvent effects in reaction kinetics and mechanisms should have had a place in Volume 2 of this treatise but as sometimes happens the author who had taken on this task failed to live up to his promise However since 1972 the physical chemistry of mixed aqueous solvents has made considerable strides so that the belated discussion of this topic by a new author is built on evidence that was not available at the time of publication of Volume 2

Concrete Solutions 2014 Michael Grantham, P A Muhammed Basheer, Bryan Magee, Marios Soutsos, 2014-08-18 The Concrete Solutions series of International Conferences on Concrete Repair began in 2003 with a conference held in St Malo France in association with INSA Rennes Subsequent conferences have seen us partnering with the University of Padua in 2009 and with TU Dresden in 2011 This conference is being held for the first time in the UK in association with Queen's University Belfast and brings together delegates from 36 countries to discuss the latest advances and technologies in concrete repair Earlier conferences were dominated by electrochemical repair but there has been an interesting shift to more unusual methods such as bacterial repair of concrete plus an increased focus on service life design aspects and modelling with debate and discussion on the best techniques and the validity of existing methods Repair of heritage structures is also growing in importance and a number of the papers have focused on the importance of getting this right so that we may preserve our rich cultural heritage of historic structures This book is an essential reference work for those working in the concrete repair field from Engineers to Architects and from Students to Clients

Effects of Context-oriented Learning on Student Interest and Achievement in Chemistry Education Sabine Fechner, 2009 Der Chemieunterricht sieht sich nicht erst seit PISA mit dem Anspruch konfrontiert einerseits die Lernenden von der Relevanz des Faches zu überzeugen und andererseits zu gewährleisten dass sie genug vernetztes Konzeptwissen erwerben um eine naturwissenschaftliche Berufslaufbahn wählen zu können An diesem Spannungsfeld setzt die vorgestellte Studie an anhand eines experimentellen Designs wird der Erwerb von Wissen in lebensweltlichen im Gegensatz zu fachlichen Kontexten evaluiert Fünf Experimentierphasen dienen als Grundlage um die Effizienz der jeweiligen Kontexte in Verbindung mit wiederholenden Vernetzungsmassnahmen zu

untersuchen Die gewählten Messinstrumente ermöglichen hierbei eine detaillierte Beschreibung der Effekte der Interventionsmassnahmen beim kooperativen Arbeiten im Chemieunterricht Especially since the weak results of German students in large scale assessments like PISA chemistry education has been challenged It is supposed to bridge the gap between making it relevant to the learner and ensuring that learners acquire enough content knowledge to cope with university demands The presented study emanates from this ambiguity by evaluating the acquisition of content knowledge in different contexts By means of an experimental design student learning gains in tasks with a real life problem situation are compared to a laboratory setting A one week intervention serves as a basis to assess the effects of such contexts in a collaborative and inquiry based learning environment The selected test instruments offer the possibility to shed light on differentiated effects of contexts with relation to learner prerequisites

Monthly Catalog of United States Government Publications ,1971 **Monthly Catalog of United States Government Publications, Cumulative Index** United States. Superintendent of Documents,1976

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