

Overview of Chemical Bonding and Chemical Reactions

Fill in the blanks with the correct information about chemical bonding.

1. An _____ is an atom that has either lost or gained an electron.
2. An atom becomes _____ only when its outermost shell is filled with electrons.
3. A chemical bond is formed when atoms _____ or _____ electrons.
4. An _____ is a force of attraction between opposing charges of the ions in the compound.
5. A _____ involves two or more electrons sharing electrons to achieve stability.
6. A _____ molecule has a positive and a negative end.
7. A non-polar molecule is one whose constituent atoms _____ electrons equally.
8. A _____ ion is a group of atoms with a positive or negative charge.
9. NaCl is an _____ compound.
10. _____ is the property that determines whether or not a molecule is polar.
11. _____ do not easily form bonds, as their outermost shells are filled.
12. The number of electrons in the outermost shell of an atom _____ as we go left to right in a period.
13. The reactivity of halogens _____ as we move down their group.
14. H_2 is a _____ covalent compound.
15. _____ bonds are formed when metals share their electrons.
16. A _____ bond has two pairs of electrons.

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