or indirectly before the examination. In addition to the fittings and reagents normally contained in a Chemistry laboratory, the following apparatus and materials will be 2. required by each candidate:

(a) one burette of 50 cm3 capacity;

(b) one pipette, either 20 cm3 or 25 cm3. All candidates at one centre must use pipettes of the same volume. These should be clean and free from grease;

(c) the usual apparatus for titration;

(d) the usual apparatus and reagents for qualitative work including the following with all reagents appropriately labelled:

dilute sodium hydroxide solution; (i)

dilute hydrochloric acid; (iii)

dilute trioxonitrate (V) acid; (iiii)

silver trioxonitrate (V) solution; (iv)

aqueous barium chloride: (v)

aqueous ammonia: (vi)

lime water: (viii)

(viii) red and blue litmus papers;

dilute tetraoxosulphate (VI) acid; (ix)

acidified potassium dichromate solution. (x)

(e) spatula:

(f) filtration apparatus;

(g) one beaker:

(h) one boiling tube;

(i) four test tubes:

(i) methyl orange indicator;

(k) glass rod:

(1) wash bottle containing distilled/deionized water;

(m) burning splint;

(n) watch glass;

(o) bunsen burner/source of heat;

(p) droppers

(q) mathematical table/calculator;

(r) phenolphthalein indicator.

2015 Waec Chemistry 3 Practical Answer

Tao Wei

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "2015 Waec Chemistry 3 Practical Answer," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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