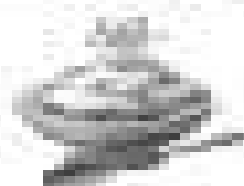


Atomic Probing Structure of the Atom

Name _____
Date _____



Purpose:

To determine which elements are present in the gas by observing the color of the emission spectrum.

Part I: Making Observations in the Atomic Probing

When the gas in the tube is excited, it emits light. The color of the light is determined by the energy levels of the atoms. The energy levels are determined by the number of protons and neutrons in the nucleus of the atom.

Observation 1: The emission spectrum of hydrogen gas is shown in the figure. The spectrum shows four distinct lines of color.

Using the color of the lines, you can determine the elements present in the gas. (Color of the lines is shown in the figure.)

Observation 2: The emission spectrum of helium gas is shown in the figure. The spectrum shows five distinct lines of color. The colors of the lines are different from the colors of the lines in the hydrogen spectrum.

From the color of the lines, you can determine the elements present in the gas. (Color of the lines is shown in the figure.)

Observation 3: The emission spectrum of neon gas is shown in the figure. The spectrum shows six distinct lines of color. The colors of the lines are different from the colors of the lines in the hydrogen and helium spectra.

From the color of the lines, you can determine the elements present in the gas. (Color of the lines is shown in the figure.)

Answers Lesson 11 Atomic Pudding

John Hudson Tiner



Answers Lesson 11 Atomic Pudding:

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Atomic and Molecular Structure Walter J. Lehmann, 1972

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, 2022-08-25 UGC NET LIFE SCIENCE unit 1 **Mega Molecules** Hans-Georg Elias, 2013-06-29 All life is based on big molecules scientifically called macromolecules Humans animals and plants cease to exist without these structural reserve and transport molecules No life can be propagated without macromolecular DNA and RNA Without macromolecules we would only dine on water sugars fats vitamins and salts but had to relinquish meat eggs cereals vegetables and fruits We would not live in houses since wood and many stones consist of macromolecules Without macromolecules no clothes since all fibers are made from macromolecules No present day car could run All tires are based on macromolecules Without macromolecules no photographic films no electronics If macromolecules are so important then why is commonly so little known about their roles and why are they so little mentioned in school if at all As often in human history tradition is important and science makes no exception Chemistry was established as the chemistry of low molecular weight compounds since these were most easy to investigate characterize and convert A beautiful tower of thought was erected by the chemical sciences long before the idea of giant molecules macromolecules took hold There was no space for newcomers in this tower Even today one can learn about chemistry without hearing a word about macromolecules **ARUN DEEP'S SELF-HELP**

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Adopting the Track of Appearance: An Emotional Symphony within **Answers Lesson 11 Atomic Pudding**

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