

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- | | |
|---------------|---|
| <u>yes</u> | 1. A tomato smells rotten. |
| <u>no</u> | 2. A drinking glass breaks into smaller pieces. |
| <u>no</u> | 3. A piece of ice melts. |
| <u>yes</u> | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u> | 5. Candle wax burns. |
| <u>yes/no</u> | 6. Molten candle wax solidifies. |
| <u>yes</u> | 7. Green leaves turn yellow and red as the seasons change. |
| <u>yes</u> | 8. Baking powder produces a gas that makes a cake rise. |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

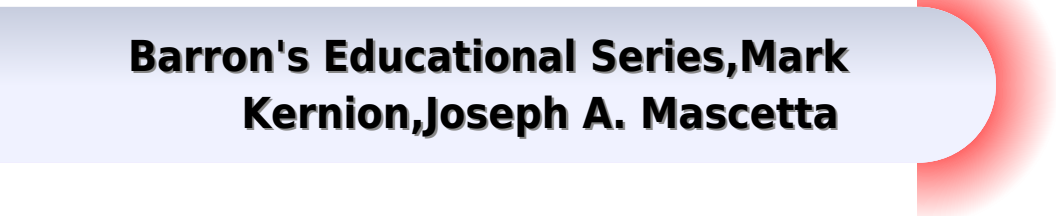
Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

Answer Key To Study Packet Chemical Reactions

**Barron's Educational Series, Mark
Kernion, Joseph A. Mascetta**



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Study Guide with Answers to Selected Problems H. Stephen Stoker, Danny V. White, 2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises

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A Life Scientist's Guide to Physical Chemistry Marc R. Roussel, 2012-04-05 Demonstrates how the tools of physical chemistry can be applied to biological questions with numerous exercises and clearly worked examples

Study Guide to Accompany Roach's Introductory Clinical Pharmacology Susan M. Ford, Sally S. Roach, 2009-10-01 This Study Guide is designed to accompany the Ninth Edition of Roach's Introductory Clinical Pharmacology

Chemistry Equations & Answers Speedy Publishing, 2014-08-26 General chemistry inorganic chemistry organic chemistry and biochemistry are all difficult courses requiring much memorization for the student Essentially there is no easy way to learn formulas and facts This is why chemistry classes are such challenges to students even the best ones However a chemistry equations and answers study guide can help the student When used as a quick reference guide it can be used often to determine the formulas needed for various questions The astute student can cleverly devise ways to make the guide useful for test questions or other circumstances requiring one of the many chemistry equations

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Golanty, Kelli McCormack Brown, 2000 Health Wellness **Spotlight Science Teacher Support Pack 7: Framework Edition** Keith Johnson, 2003 This Framework Edition Teacher Support Pack offers comprehensive support and guidance providing the best possible learning experience for your students and saving time for everyone in the department *Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e* Kenneth W.

Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond s text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry **Spotlight**

Science Teacher Support Pack 9 Keith Johnson, Lawrie Ryan, Sue Adamson, 2004 This Framework Edition Teacher Support Pack offers support and guidance *Chemical Reactions, Materials and Particles* Gill Murphy, 2006-03

Specials are a teaching resource designed for KS3 students whose literacy skills are considerably lower than their age These books have an older format to counteract the simple text and cover various topics They include activities visuals and assessment sheets as well as teacher pages *Fundamentals of Ordinary Differential Equations* Mohit Chatterjee, 2025-02-20

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Study Guide with Selected Solutions Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

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