

Name _____

BONDING



Directions: Use the chart below and your periodic table to answer the questions that follow.

PART I: COMPARING ELEMENTS

Group One and Group Two both show different compounds made by bonding different elements together. There are two types of bonding in food science, as shown in these two groupings. Let's take a closer look at the elements in each compound to see what makes a compound IONIC or COVALENT.

IONIC BONDS	COVALENT BONDS
NaCl (Sodium Chloride)	H ₂ O (Dihydrogen Monoxide)
MgO (Magnesium Oxide)	CH ₄ (Methane)
Li ₂ O (Lithium Oxide)	CO ₂ (Carbon Dioxide)
KF (Potassium Fluoride)	HF (Hydrofluoric Acid)
FeBr ₃ (Iron III Bromide)	NH ₃ (Ammonia)
CaCl ₂ (Calcium Chloride)	NO ₂ (Nitrogen Dioxide)
NiI ₂ (Nickel Iodide)	C ₆ H ₁₂ O ₆ (Glucose)
BaS (Barium Sulfide)	CF ₄ (Carbon Tetrafluoride)

1. List out the FIRST element in each of the GROUP ONE compounds.
2. What do you notice about the FIRST element that makes up each compound in GROUP ONE?
3. What side of the periodic table do you find these FIRST elements on?
4. List out the SECOND element in each of the GROUP ONE compounds.
5. What do you notice about the SECOND element that makes up each compound in GROUP ONE?
6. What side of the periodic table do you find these SECOND elements on?

BIG IDEA #1: When a _____ and a _____ come together, they make an IONIC BOND.

7. What do you notice about ALL of the elements in GROUP TWO?
8. What side of the periodic table do you find ALL of these elements on?

Answers Chemical Bonding Pogil

TD Snyder

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Answers Chemical Bonding Pogil

1. Understanding the eBook Answers Chemical Bonding Pogil
 - The Rise of Digital Reading Answers Chemical Bonding Pogil
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers Chemical Bonding Pogil
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers Chemical Bonding Pogil
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers Chemical Bonding Pogil
 - Personalized Recommendations
 - Answers Chemical Bonding Pogil User Reviews and Ratings

- Answers Chemical Bonding Pogil and Bestseller Lists
- 5. Accessing Answers Chemical Bonding Pogil Free and Paid eBooks
 - Answers Chemical Bonding Pogil Public Domain eBooks
 - Answers Chemical Bonding Pogil eBook Subscription Services
 - Answers Chemical Bonding Pogil Budget-Friendly Options
- 6. Navigating Answers Chemical Bonding Pogil eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers Chemical Bonding Pogil Compatibility with Devices
 - Answers Chemical Bonding Pogil Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers Chemical Bonding Pogil
 - Highlighting and Note-Taking Answers Chemical Bonding Pogil
 - Interactive Elements Answers Chemical Bonding Pogil
- 8. Staying Engaged with Answers Chemical Bonding Pogil
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers Chemical Bonding Pogil
- 9. Balancing eBooks and Physical Books Answers Chemical Bonding Pogil
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers Chemical Bonding Pogil
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers Chemical Bonding Pogil
 - Setting Reading Goals Answers Chemical Bonding Pogil
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers Chemical Bonding Pogil
 - Fact-Checking eBook Content of Answers Chemical Bonding Pogil
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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