

PHYSICAL: Kinetic Molecular Theory

Learning Objectives

- At Standard-Hardest difficulty, we ask that you comprehend the Kinetic Theory, and use mathematical modeling to understand the relationship of the kinetic molecular theory to Standard-Hardest difficulty scientific theories.
- At Standard-Hardest difficulty, we ask that you comprehend the Kinetic Theory, and use mathematical modeling to understand the relationship of the Kinetic Theory to Standard-Hardest difficulty scientific theories.

Model 1: Representation of States in Different Phases

The following figure illustrates the representation of the three states of matter: solid, liquid, and gas.

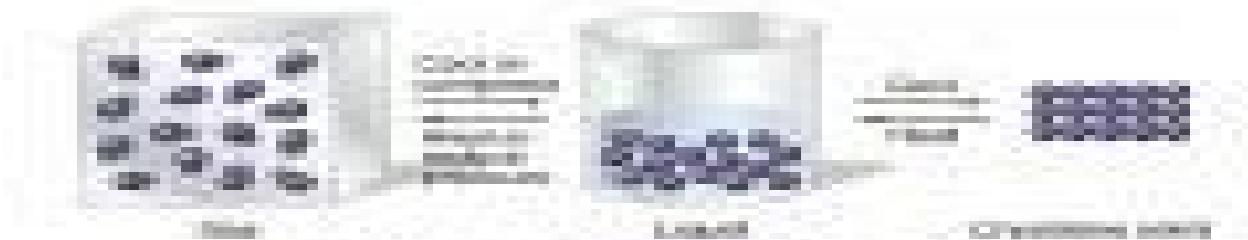


Figure 1: Representation of States in Different Phases

Model 1: Representation of States in Different Phases. The figure illustrates the three states of matter: solid, liquid, and gas. The particles are represented by small circles. In the solid state, the particles are in a fixed, regular arrangement. In the liquid state, the particles are in a disordered, close-packed arrangement. In the gaseous state, the particles are in a disordered, widely spaced arrangement.

Key Concepts

1. Model 1: Representation of States in Different Phases. The figure illustrates the three states of matter: solid, liquid, and gas. The particles are represented by small circles. In the solid state, the particles are in a fixed, regular arrangement. In the liquid state, the particles are in a disordered, close-packed arrangement. In the gaseous state, the particles are in a disordered, widely spaced arrangement.

Figure 1: Representation of States in Different Phases

	SOLID	LIQUID	GAS
Particle Arrangement	Particles are in a fixed, regular arrangement. They are in contact with each other.	Particles are in a disordered, close-packed arrangement. They are in contact with each other.	Particles are in a disordered, widely spaced arrangement. They are not in contact with each other.
Particle Movement	Particles are in a fixed position and can only vibrate. They have low kinetic energy.	Particles are in a disordered position and can move. They have moderate kinetic energy.	Particles are in a disordered position and can move. They have high kinetic energy.

Answers Kinetic Molecular Theory Pogil

SJ Ball



Answers Kinetic Molecular Theory Pogil:

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