

CHEMISTRY – COURSEWORK EXAM

MULTIPLE CHOICE ANSWERS

Questions 1-16

16) During the electrolysis of aqueous copper (II) sulphate solution using inert electrodes, the ions migrating to the cathode are:

- (A) $\text{Cu}^{2+}(\text{aq})$ and $(\text{aq}) \text{SO}_4^{2-}$
- (B) $\text{H}^+(\text{aq})$ and $\text{OH}^-(\text{aq})$
- (C) $\text{SO}_4^{2-}(\text{aq})$ and $\text{OH}^-(\text{aq})$
- (D) $\text{Cu}^{2+}(\text{aq})$ and $\text{H}^+(\text{aq})$

ANSWER: (D)

17) What quantity of electricity passes through the electrolyte when a current of 100 A flows for 10 minutes?

- A) 100 C
- B) 600 C
- C) 60 000 C

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AW Rasmussen



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web goal the goal of this module is to introduce osha to the participants starting with the formation and history of osha objectives by the conclusion of module 1 participants will be able to describe the meaning of osha give the history of osha including the major events leading to the formation of osha describe what osha's mission is

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