

7. Alkanols can be manufactured from alkenes by the initial reaction of alkenes with
- bromine in tetrachloromethane.
 - concentrated tetraoxosulphate(VI) acid.
 - aqueous potassium tetraoxomanganate(VII).
 - sodium hydroxide solution.
8. Which of the following statements about the standard hydrogen electrode is **not** correct?
- The hydrogen gas is at a pressure of 1 atmosphere
 - A solution containing 1 mol dm^{-3} of $\text{H}^+_{(\text{aq})}$ ions is used
 - A platinum electrode is used
 - The temperature is kept at 20°C
9. If 60 g of M combines with 24 g of oxygen, what would the empirical formula of the oxide be? [O = 16.0, M = 120]
- MO
 - MO₂
 - MO₃
 - MO₄
10. The products of the electrolysis of dilute sodium chloride using carbon electrodes are
- | | Anode | Cathode |
|----|----------|-----------|
| A. | chlorine | sodium. |
| B. | oxygen | hydrogen. |
| C. | chlorine | hydrogen. |
| D. | hydrogen | oxygen. |
11. Determine the quantity of electricity used when a current of 0.20 amperes is passed through an electrolytic cell for 60 minutes.
- 12 C
 - 120 C
 - 360 C
 - 720 C
12. Oxochlorate(I) acid is used as a bleaching agent because it is
- a weak acid.
 - a reducing agent.
 - an oxidizing agent.
 - a strong acid.
13. The IUPAC name for $\text{CH}_3\text{CHCH}_3\text{CHClCHCH}_3\text{CH}_2\text{CH}_3$ is
- 2, 4 - dimethyl - 3 - chlorohexane.
 - 3, 5 - dimethyl - 4 - chlorohexane.
 - 4 - chloro - 3, 5 - dimethylhexane.
 - 3 - chloro - 2, 4 - dimethylhexane.
14. A colourless gas with a pungent smell is evolved when dilute hydrochloric acid is added to a sample of a salt. The gas evolved could turn
- acidified $\text{K}_2\text{Cr}_2\text{O}_7$ solution colourless.
 - acidified KMnO_4 solution colourless.
 - $\text{Fe}(\text{NO}_3)_3$ solution green.
 - $\text{Pb}(\text{NO}_3)_2$ paper black.
15. If 5.0 g of marble reacts with 25.0 cm^3 hydrochloric acid, which of the following combinations has the fastest reaction rate?
- Marble chips and $2.0 \text{ mol dm}^{-3} \text{ HCl}_{(\text{aq})}$
 - Marble chips and $2.5 \text{ mol dm}^{-3} \text{ HCl}_{(\text{aq})}$
 - Powdered marble and $2.5 \text{ mol dm}^{-3} \text{ HCl}_{(\text{aq})}$
 - Powdered marble and $2.0 \text{ mol dm}^{-3} \text{ HCl}_{(\text{aq})}$

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