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Albert A Gayle



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web for example take the reaction of lead ii nitrate with hydrochloric acid to form lead ii chloride and nitric acid shown below
$$\text{Pb(NO}_3)_2(\text{aq}) + 2\text{HCl}(\text{aq}) \rightarrow \text{PbCl}_2(\text{s}) + 2\text{HNO}_3$$

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web the net ionic equation for equation 1 is given as equation 3 $\text{s}^{2-} \text{ aq} + \text{Fe}^{2+} \text{ aq} \rightarrow \text{FeS} \text{ s}$ eq 3 8 note that the phase indicator aq is optional on ionic species but is

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Hg_2^{2+} and Sr^{2+} the net ionic equation is

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