Name : _______ Date: ______

Classifying Chemical Reactions

- A. Classify the following reaction as synthesis, decomposition, single-replacement, double-replacement, and combustion:
- 1) Al₂(SO₄)₃ + 3 BaCl₂ → 3 BaSO₄ + 2 AlCl₃ _____
- 2) Al₂S₃ 2 Al + 2 S
- 3) 2 NaOH + CuSO₄ → Na₂SO₄ + Cu(OH)₂ _____
- 4) Fe + H₂SO₄ → FeSO₄ + H₂ _____
- 5) C₄H₁₂ + 7O₂ 4CO₂ + 6H₂O _____
- B. Balance the following equations and indicate the type of reaction:

1)
$$H_2S$$
 + O_2 \longrightarrow SO_2 + H_2O _____

Answers For Classifying Chemical Reactions

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in multi disciplinary cooperation with bio chemical engineers by providing an overview of this exciting field The textbook is organized into seven distinct parts Structure of the chemical industry and bio chemical processes Bio Chemical reaction engineering Molecular separations distillation extraction absorption adsorption Mechanical separations filtration sedimentation membranes Particle and final product manufacturing Development scale up design and safety of processes Major industrial process descriptions

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