$$2 \text{ KClO}_2(s) \xrightarrow{\Delta} 2 \text{ KCl}(s) + 3 \text{ O}_2(g)$$

$$MnO_2$$

According to the equation above, how many moles of potassium chlorate, KCIO₁, must be decomposed to generate 1.0 L of O₂ gas at standard temperature and pressure?

- A) $\frac{1}{3} \left[\frac{1}{22.4} \right]$ mol
- $89 \quad \frac{1}{2} \left(\frac{1}{22.4} \right) \text{ mol}$
- C) $\frac{2}{3} \left[\frac{1}{22.4} \right]$ med
- D) $\frac{3}{2} \left(\frac{1}{22.4} \right)$ mel
- E) $2\left[\frac{1}{22.4}\right]$ mod
- Which of the following ions has the same number of electrons as Br.?
 - 45 300
 - B6 K1
 - C) Se**
 - 100
 - E) Cl
- In solid methans, the forces between neighboring CH, molecules are best characterized as
 - A) lionic bonds.
 - B) covulent bonds
 - C) hydrogen bonds
 - D) ion-dipole forces
 - E) London (dispersion) forces

- 21. Of the following electron configurations of neutral atoms, which represents an atom in an excited state?
 - A) Lettertheir
 - Letterthethethet
 - C) Lettertpytet
 - D) Le'Ga'Qp'Sa'Bp'
 - Leitzeitzphiteitige
- 22. Which of the following is a neepolar molecule that contains polar bonds?
 - At Fo
 - B) CHF,
 - C) CO₀
 - Di HC1
 - E) NH-
- The exidation state that is common to aqueous ions of Fe. Mn. and Zn is
 - 465 141
 - B0 +2
 - CN +43.
 - Di +4
 - D +5
- Which of the following shows the correct number of protons, neutrons, and electrons in a neutral cestum-134 atom?

	Protons	Neutrons	Electrons
Alt	5.5	55	55
800	5.5	79	55
623	3.5	79	79
10)	39	33	79
203	134	55	134

Ap Chemistry Multiple Choice Exam

Neil D. Jespersen, Pamela Kerrigan

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