

Question 1: Water at 20°C falls from a height of 854 m. If the whole energy is used in increasing the temperature, find out the final temperature.
Specific heat of water is 4200 J/K kg .

Given that

$$T_1 = 20^{\circ}\text{C} = 20 + 273 = 293 \text{ K}$$

$$\text{height} = h = 854 \text{ m}$$

$$\text{Specific heat of water} = 4200 \text{ J K}^{-1} \text{ kg}^{-1}$$

$$\text{Final temperature} = T_2 = ?$$

$$\text{P.E} = mgh \quad \text{--- (1)}$$

Solution:

We know that

$$\Delta Q = cm \Delta T \quad \text{--- (2)}$$

Whole energy is used in increasing the temperature (Given)

$$\Delta Q = \text{P.E}$$

From (1) and (2)

$$cm \Delta T = mgh$$

$$c \Delta T = gh$$

$$c (T_2 - T_1) = gh$$

$$T_2 - T_1 = \frac{gh}{c}$$

$$T_2 = \frac{gh}{c} + T_1$$

$$= \frac{9.8 \times 854}{4200} + 293$$

$$= \frac{8369.2}{4200} + 293 = 1.99 + 293$$

$$T_2 = 294.99 \text{ K} = 295 - 273^{\circ}\text{C}$$

$$T_2 = 22^{\circ}\text{C}$$

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Theoretical Chemistry from the Standpoint of Avogadro's Rule & Thermodynamics Walther Nernst, 1895 Ice and Refrigeration, 1892

Gravity and the Quantum Jasjeet Singh Bagla, Sunu Engineer, 2017-02-23 This book provides a compilation of in depth articles and reviews on key topics within gravitation cosmology and related issues It is a celebratory volume dedicated to Prof Thanu Padmanabhan Paddy the renowned relativist and cosmologist from IUCAA India on the occasion of his 60th birthday The authors many of them leaders of their fields are all colleagues collaborators and former students of Paddy who have worked with him over a research career spanning more than four

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Provides an essential treatment of the subject and rigorous methods to solve all kinds of energy engineering problems

Mere Thermodynamics Don S. Lemons,2008-12-08 An accessible exploration of a notoriously difficult subject this gem of physics pedagogy should be required reading for all courses on thermodynamics Robert C Hilborn University of Texas at Dallas Presenting classical thermodynamics as a concise and discrete whole Mere Thermodynamics is a perfect tool for teaching the theory s central concepts and methods The book features end of chapter practice problems an appendix of worked problems a glossary of terms and an annotated bibliography Accomplished teacher Don S Lemons uses thermodynamics to solve problems from a broad range of physics He illustrates not only the fundamentals of the subject but also advanced topics such as the relationship between the second law of thermodynamics and entropy He highlights the intellectual structure and history of the discipline and explores the logical consequences of each of the famous three laws Lemons explains and develops the first two laws and their corollaries the methods and applications of thermodynamics and the third law as well as non fluid variables equilibrium and stability and two phase systems Thermodynamics of the

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Lee,2012-02-28 This book is the expanded edition of the first book entitled Chemical Thermodynamics for Metals and Materials This new version presents thermodynamics of materials with emphasis on the chemical approach and is thus suitable for students in materials science and metallurgical engineering as well as related fields such as chemical

engineering and physical chemistry

Biothermodynamics Part A Michael L. Johnson, Jo M. Holt, Gary K.

Ackers, 2009-03-14 In the past several years there has been an explosion in the ability of biologists molecular biologists and biochemists to collect vast amounts of data on their systems This volume presents sophisticated methods for estimating the thermodynamic parameters of specific protein protein protein DNA and small molecule interactions The use of thermodynamics in biological research is used as an energy book keeping system While the structure and function of a molecule is important it is equally important to know what drives the energy force These methods look to answer What are the sources of energy that drive the function Which of the pathways are of biological significance As the base of macromolecular structures continues to expand through powerful techniques of molecular biology such as X ray crystal data and spectroscopy methods the importance of tested and reliable methods for answering these questions will continue to expand as well

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Theory of Nonequilibrium Ensembles, Irreversible Thermodynamics, and Generalized Hydrodynamics Byung Chan Eu, 2016-08-02 This book presents the fundamentals of irreversible thermodynamics for nonlinear transport processes in gases and liquids as well as for generalized hydrodynamics extending the classical hydrodynamics of Navier Stokes Fourier and Fick Together with its companion volume on relativistic theories it provides a comprehensive picture of the kinetic theory formulated from the viewpoint of nonequilibrium ensembles in both nonrelativistic and in Vol 2 relativistic contexts Theories of macroscopic irreversible processes must strictly conform to the thermodynamic laws at every step and in all approximations that enter their derivation from the mechanical principles Upholding this as the inviolable tenet the author develops theories of irreversible transport processes in fluids gases or liquids on the basis of irreversible kinetic equations satisfying the H theorem They apply regardless of whether the processes are near to or far removed from equilibrium or whether they are linear or nonlinear with respect to macroscopic fluxes or thermodynamic forces Both irreversible Boltzmann and generalized Boltzmann equations are used for deriving theories of irreversible transport equations and generalized hydrodynamic equations which rigorously conform to the tenet All observables described by the so formulated theories therefore also strictly obey the tenet *Chemical Thermodynamics For Metals And Materials (With Cd-rom For Computer-aided Learning)* Hae-geon Lee, 1999-10-13 A number of thermodynamic books claiming to be original in both presentation and approach have been published However thermodynamics is still a confusing subject for uninitiated students and an easy to forget one for graduate engineers In order to solve these problems this computer aided learning package textbook and CD ROM takes a new approach This package is unique and beneficial in that it simulates a classroom lecture it actually writes important equations and concepts on a virtual board underlines draws circles places ticks to emphasise important points draws arrows to indicate relationships uses colours for visual effect erases some parts to write new lines and even repeats some parts of the lesson to stress their importance This realistic simulation is made possible by the employment of the multimedia capabilities of the modern day computer Readers are not just passively presented with thermodynamics they can also interactively select and repeat any particular topic of interest as many times as they want This flexibility allows readers to choose their own pace of presentation This complementary set is in many important respects better than the books that are currently available on the subject **Analytical Thermodynamics** Dongqing Li, 2024-12-07 This second edition presents an enriched and expanded exploration of the fundamental principles of thermodynamics tailored for graduate level studies Drawing on over three decades of academic teaching experience the author has refined the content making it more accessible and comprehensive Chapter 1 has been restructured for clarity delineating Legendre Transformation and Thermodynamic Potentials into separate sections while the treatment of Chemical Potentials has been significantly augmented encompassing two component ideal gas mixtures and a re derivation of chemical potentials for dilute solutions Additionally the section on thermodynamic stability now boasts enhanced explanations and illustrative figures

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web wed may 1st 2024 school principals day organize an event or write a card to show your appreciation for the educators that run our schools school principals work hard to make kids lives and educations better have we missed something you can have great teachers but if you don t have a good principal you won t have a good school eli broad

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web celebrate school principals day on may 1 and show appreciation to the amazing school administrators who work hard every day to create an enriching learning environment for students national school principals day was established as a way of honoring our nation s school leaders