Chemical Formulas and Chemical Compounds

SECTION 1

SHORT ANSWER Answer the following questions in the space provided.

- In a Stock system name such as iron(III) sulfate, the Roman numeral tells us
 - (a) how many atoms of Fe are in one formula unit.
 - (b) how many sulfate ions can be attached to the iron atom.
 - (c) the charge on each Fe ion.
 - (d) the total positive charge of the formula unit.
- Changing a subscript in a correctly written chemical formula
 - (a) changes the number of moles represented by the formula.
 - (b) changes the charges on the other ions in the compound.
 - (c) changes the formula so that it no longer represents the compound it previously represented.
 - (d) has no effect on the formula.
- The explosive TNT has the molecular formula C₂H₂(NO₂)₃.
 - a. How many elements make up this compound?
 - b. How many oxygen atoms are present in one molecule of C₂H₃(NO₂)₃?
 - c. How many atoms in total are present in one molecule of C₇H₃(NO₂)₃?
 - d. How many atoms are present in a sample of 2.0 × 10²³ molecules of C₇H₅(NO₂)₃?
- 4. How many atoms are present in each of these formula units?
 - a. Ca(HCO₃)₂
 - b. C₁₂H₂₂O₁₁
 - c. Fe(ClO₂)₃
 - d. Fe(ClO₃)₂
- 5. ______ a. What is the formula for the compound dinitrogen pentoxide?
 - b. What is the Stock system name for the compound FeO?
 - c. What is the formula for sulfurous acid?
 - d. What is the name for the acid H₃PO₄?

Answers Review Chemical Formulas And Chemical Compounds

James E. Brady, Fred Senese

Answers Review Chemical Formulas And Chemical Compounds:

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that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement **Hazmat Chemistry Study Guide (Second Edition)** Jill Meryl Levy,2011

This book delves into Answers Review Chemical Formulas And Chemical Compounds. Answers Review Chemical Formulas And Chemical Compounds is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Answers Review Chemical Formulas And Chemical Compounds, encompassing both the fundamentals and more intricate discussions.

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- 4. In chapter 3, the author will examine the practical applications of Answers Review Chemical Formulas And Chemical Compounds in daily life. This chapter will showcase real-world examples of how Answers Review Chemical Formulas And Chemical Compounds can be effectively utilized in everyday scenarios.
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 The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly
 - recommended for anyone seeking to gain a comprehensive understanding of Answers Review Chemical Formulas And Chemical Compounds.

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