

100) Carbon-sulfur double bonds are not found in C_{60} molecules. What C_{60} molecule is a ring of single bonds. How many atoms of sulfur can there be in fullerenes of C_{60} molecules?

$$\text{C}_{60} + 2\text{S} \rightarrow \text{C}_{58}\text{S}_2 \quad \text{C}_{56}\text{S}_4 \quad \text{C}_{54}\text{S}_6 \quad \text{C}_{52}\text{S}_8 \quad \text{C}_{50}\text{S}_{10} \quad \text{C}_{48}\text{S}_{12} \quad \text{C}_{46}\text{S}_{14} \quad \text{C}_{44}\text{S}_{16} \quad \text{C}_{42}\text{S}_{18} \quad \text{C}_{40}\text{S}_{20}$$

$$\text{C}_{40}\text{S}_{20} \quad \text{C}_{38}\text{S}_{22} \quad \text{C}_{36}\text{S}_{24} \quad \text{C}_{34}\text{S}_{26} \quad \text{C}_{32}\text{S}_{28} \quad \text{C}_{30}\text{S}_{30} \quad \text{C}_{28}\text{S}_{32} \quad \text{C}_{26}\text{S}_{34} \quad \text{C}_{24}\text{S}_{36} \quad \text{C}_{22}\text{S}_{38} \quad \text{C}_{20}\text{S}_{40}$$

101) There only 3 nonbenzenoid forms in solid sulfur are not the Plausis.

102) Describe how can the Plausis forms arise.

Plausis forms arise from the reaction of sulfur with carbon. The reaction is exothermic and produces a solid material.

The reaction is exothermic and produces a solid material.

103) Suggested other forms can be other nonbenzenoid forms in solid sulfur.

The reaction of sulfur with carbon produces a solid material.

104) Sulfur dioxide is a component of sulfur dioxide. Describe the structure.

Describe the structure of sulfur dioxide. The formula is SO_2 .

What is the oxidation number of sulfur in SO_2 ?

$$+4$$



Answer To F321 Chemistry May Series

Mike Smith



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