

QUESTION 9 (contd.)

9d) Just before boiling

Heat supplied = Heat gained by water

$$Vt = mc\Delta\theta$$

$$2000 \times (2 \times 60) = m(4200) \times (100 - 20)^{\circ}\text{C}$$

$$\text{mass of water} = \frac{2000 \times 120}{4200 \times 80}$$

$$m = \frac{5}{7} \text{ kg}$$

During Vaporisation,

Heat supplied = Heat absorbed

$$Vt = mL$$

$$2000 \times (6 \times 60) = \left(\frac{45}{100}\right)m \times L$$

$$720000 = \left(\frac{45}{100} \times \frac{5}{7}\right)L = \frac{9}{28}L$$

$$L = \frac{28}{9}(720000)$$

Latent heat of vaporization, $L = 2240000$

$$L = 2.24 \times 10^6 \text{ J/kg}$$

(ii) It is assumed that no heat was lost to the surrounding through conduction or convection

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