[1]

[Score]

- (23) The volume of a definite amount of hydrogen gas taken at 2 atm pressure and 25°C temperature is 400 mL.
 - a) Suggest a method to increase the volume of this gas without change in pressure.
 - b) If the volume of the gas is changed to 200 mL, what will be the new pressure?

 (Hint: Temperature is constant)

 [2]
 - c) State the gas law used to solve the problem (b). [1]
- Q4) Sugar taken in a watch glass was found to be charred by adding a substance 'X'.
 - a) What is the substance 'X'? [1]
 - b) What is the reason for charring of sugar? [1]
 - e) Which catalyst is used in the industrial preparation of the substance 'X'?

OR.

Chemical equation of the industrial preparation of SO3 is given.

2 SO₂ + O₂ ===== 2SO₃ + Heat

What is the influence of the following factors in this system at equilibrium.

- a) Increasing the amount of O₂. [1]
- b) SO₃ is removed from the system. [1]
- c) Decreasing the temperature. [1]

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