CHM 111 Analysis of Aluminum-Zinc Alloy using Ideal Gas Law

Report Sheet

Data Sheet:

	Trial 1	Trial 2	
Mass of gelatin capsule	1.7382 g		_ g
Mass of alloy sample plus capsule	1.9852 g		_ g
Mass of empty beaker	104.3024 g		g
Mass of beaker plus displaced water	346.225g		g
Barometric pressure		m Hg	
Temperature	_ 24	_°C	
Calculations Mass of alloy sample	0.247 g		_ g
Mass of displaced water	241.9226 g	1	g
Volume of displaced water	mL		mL
Volume of H ₂ , V	L		L
Temperature of H ₂ , T	297.15	_ K	
Vapor pressure of water at T	22.4	_ mm Hg	
Pressure of dry H2,	755.7 mm Hg;	0.9943 atm	
Moles H2 from sample,	moles		moles
Moles H2 per gram of sample,	moles/g		moles/g %
Al (read from graph)	%	m Hg	_%
Unknown no	THE RESERVE OF THE PERSON OF T		TH

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