

Chemistry, Part 1

Module 14 Notes



Big Question

How are metals used in chemical reactions? and produced by chemical reactions?

Big Idea

Using chemical analysis, you can identify the metals of a substance or pure metal and other substances. A metal is often used in manufacturing. It is used in construction or industrial analysis with chemical reactions.

Learning Objectives

After completing this module, students will be able to:

- Use the law of conservation of mass to balance chemical reactions.
- Use chemical reactions to describe processes of materials.
- Compare different metals and reactions.
- Describe the uses of different metals.

Module 14 Objectives

Learning Objective Write and balance chemical equations.

Fill in the blanks.

1. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (Balanced)
2. $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
3. $\text{K}_2\text{MnO}_4 + \text{HCl} \rightarrow \text{H}_2\text{MnO}_4 + \text{KCl}$

Chemical Reactions and Equations

Write the chemical equation for the reaction of iron and oxygen.

Chemical equation: $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$

Chemical equation: $2\text{Fe} + \text{O}_2 \rightarrow 2\text{FeO}$

Chemical equation: $2\text{Fe} + \text{O}_2 \rightarrow 2\text{FeO}$

Chemical equation: $2\text{Fe} + \text{O}_2 \rightarrow 2\text{FeO}$

Module 14

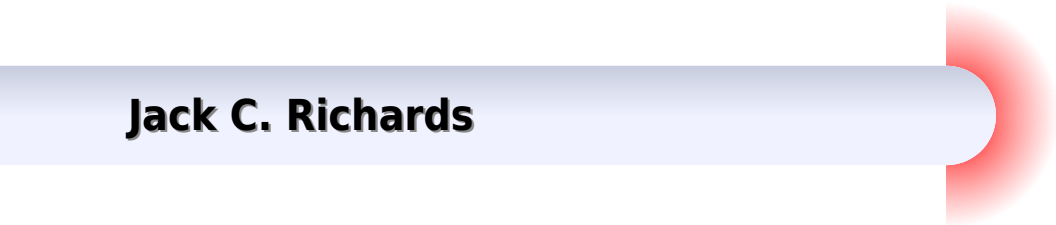
Learning Objective Write and balance chemical equations.

Fill in the blanks.

1. 1 mole of iron reacts with 1 mole of oxygen to form 1 mole of iron(II) oxide.
2. The mass of iron (Fe) is 55.85 g/mol. The mass of oxygen (O) is 16.00 g/mol. The mass of iron(II) oxide (FeO) is 71.85 g/mol.

Answers For Chemistry Module 14

Jack C. Richards



Answers For Chemistry Module 14:

Answers For Chemistry Module 14 Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has are more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **Answers For Chemistry Module 14**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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