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CHAPTER 8 STUDY GUIDE

Covalent Bonding

Section 8.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	pi bond			
When sharing of el-	ectrons occurs, the attac	hment between atom	s that results is called				
	. When suc						
released, and the proce	rss is (2)	. When t	wo or more atoms bone	d by means of			
electron sharing, the re	sulting particle is called	i a(n) (3)	3) If the electrons shared				
are centered between the two atoms, the attachment		iment is called a(n) (4	E)	If the			
sharing involves the or	verlap of parallel orbital	s, the attachment is c	alled a(n) (5)				
	about single and multip e choice that best comp			om.			
6. In what form do e	lements such as hydrog	en, nitrogen, and oxy	gen normally occur?				
 a. as single atom 				rules containing three atoms			
 b. as molecules c 	containing two atoms	d. as moi	 d. as molecules containing four atoms 				
7. How many electro	ons are shared in a doub	de covalent bond?					
a. none	b. one	c. two	d, f	our			
8. Bond length is the	e distance between						
 a. two molecules 	s of the same substance.	c. the nu	 c. the nuclei of two attached atoms. 				
b. the electrons i	n two attached atoms.	d. the orl	 the orbitals of two attached atoms. 				
9. Which of the follo	owing relationships rela-	ting to bond length is	generally correct?				
a. the shorter the	bond, the stronger the	bond					
b. the shorter the	bond, the weaker the b	ond					
 c. the shorter the 	bond, the fewer the ele	ectrons in it					
d. the shorter the	bond, the lower the bor	nd dissociation energ	y				

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Mark J. Winter

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