

EXPERIMENT 1: PERCENT SUGAR IN BUBBLE GUM

Data Sheet

Table 2: Sugar (Sucrose) Data

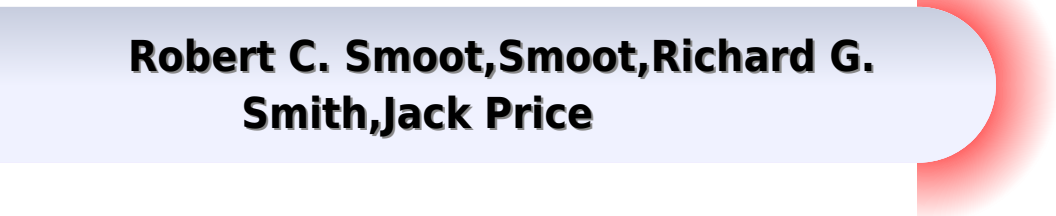
Mass of unchewed gum (g)	Mass of chewed gum (g)	Mass of sugar (g)	Percent of sugar in gum
5.6	2.6	3	46.42

Post-Lab Questions

1. What is the percent sugar? Show your calculations. $2.6/5.6 \times 100 = 46.42$
2. What is the molar mass of sugar (sucrose, $C_{12}H_{22}O_{11}$)? Show your calculations. |
3. Convert the mass of dissolved sugar to moles. Show your calculations. |
4. Convert the moles of sugar to molecules of sugar. Show your calculations.
5. Imagine that you used four pieces of gum instead of two. Calculate the number of moles. Show your calculations.
6. How does the mass of the saliva alter the experimental data?

Answer Percent Sugar In Bubble Gum Lab

**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**



Answer Percent Sugar In Bubble Gum Lab:

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **BOUDICA NARAYAN**
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Answer Percent Sugar In Bubble Gum Lab Introduction

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