8 Methane and sream react in the presence of a catalyst.

0.5 mol of methane reacts completely with 0.5 mol of steam.

What is the volume of carbon monoxide and hydrogen produced, measured at room temperature and pressure?

	of CO/dm ³	volume of H ₂ /dm ³
Α.	0.5	1.5
8	1.0	3.0
C	12.0	12.0
(D)	12.0	36.0

1 0.0	<u></u>	
	y n 2n	
	0.5 (24. 134)	
	1.5×34 = 36	

9 A compound of element X has the formula X₂O and a relative formula mass of 144.

What is element X?

- (A) copper, Cu
- B gadolinium, Gd

D. tellurium, Tel

C sulfur, S

5 4

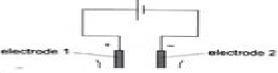
194 - 167 13%

$$\frac{123}{2} - X = 64$$

13% = X2

5

10 The diagram shows the electrolysis of concentrated hydrochloric acid and concentrated aqueous sodium chloride using carbon electrodes.





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