

4

- 8 Methane and steam react in the presence of a catalyst.



0.5 mol of methane reacts completely with 0.5 mol of steam.

What is the volume of carbon monoxide and hydrogen produced, measured at room temperature and pressure?

	volume of CO/dm ³	volume of H ₂ /dm ³
A	0.5	1.5
B	1.0	3.0
C	12.0	12.0
D	12.0	36.0

$$n = \frac{m}{M}$$

$$n = \frac{V}{24}$$

$$0.5 \times 24 = 12 \text{ dm}^3$$

$$1.5 \times 24 = 36$$

- 9 A compound of element X has the formula X_2O and a relative formula mass of 144.

What is element X?

- A copper, Cu
B gadolinium, Gd
C sulfur, S
D tellurium, Te

1 -

or

14

$$144 - 16 = 128$$

$$128 = \text{X}_2$$

$$\frac{128}{2} = \text{X} = 64$$

5

- 10 The diagram shows the electrolysis of concentrated hydrochloric acid and concentrated aqueous sodium chloride using carbon electrodes.



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Hugh Brogan



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1. Understanding the eBook Aice Chemistry May June 2013
 - The Rise of Digital Reading Aice Chemistry May June 2013
 - Advantages of eBooks Over Traditional Books
2. Identifying Aice Chemistry May June 2013
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Aice Chemistry May June 2013
 - User-Friendly Interface
4. Exploring eBook Recommendations from Aice Chemistry May June 2013
 - Personalized Recommendations
 - Aice Chemistry May June 2013 User Reviews and Ratings
 - Aice Chemistry May June 2013 and Bestseller Lists
5. Accessing Aice Chemistry May June 2013 Free and Paid eBooks
 - Aice Chemistry May June 2013 Public Domain eBooks
 - Aice Chemistry May June 2013 eBook Subscription Services
 - Aice Chemistry May June 2013 Budget-Friendly Options

6. Navigating Aice Chemistry May June 2013 eBook Formats
 - ePub, PDF, MOBI, and More
 - Aice Chemistry May June 2013 Compatibility with Devices
 - Aice Chemistry May June 2013 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Aice Chemistry May June 2013
 - Highlighting and Note-Taking Aice Chemistry May June 2013
 - Interactive Elements Aice Chemistry May June 2013
8. Staying Engaged with Aice Chemistry May June 2013
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Aice Chemistry May June 2013
9. Balancing eBooks and Physical Books Aice Chemistry May June 2013
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Aice Chemistry May June 2013
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Aice Chemistry May June 2013
 - Setting Reading Goals Aice Chemistry May June 2013
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Aice Chemistry May June 2013
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 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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